

Asymmetric Synthesis of Alcohols with Two Chiral Centres from a Racemic Aldehyde by the Selective Addition of Dialkylzinc Reagents using (1*S*,2*R*)-(–)-*N,N*-Dibutylnorephedrine and (1*S*)-(+)-Diphenyl-(1-methylpyrrolidin-2-yl)methanol as Chiral Catalysts

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Optically active alcohols with two chiral centres were obtained in up to 93% enantiomeric excess by the selective addition of dialkylzinc reagents to the racemic aldehyde, 2-phenylpropanal, using the title compounds as chiral catalysts.

Increasing interest has centred on catalytic asymmetric carbon-carbon bond-forming reactions.¹ Addition of dialkylzinc reagents to aldehydes is usually very slow, but β -amino-alcohol derivatives catalyse the addition of diethylzinc to benzaldehyde.² Although we³ and others⁴ have reported the

enantioselective addition of dialkylzinc reagents to aldehydes, the structures of the alcohols prepared have been limited to those with a single chiral centre. In connection with the synthesis of alcohols with two chiral centres, the diastereoselective addition of organometallic reagents to racemic

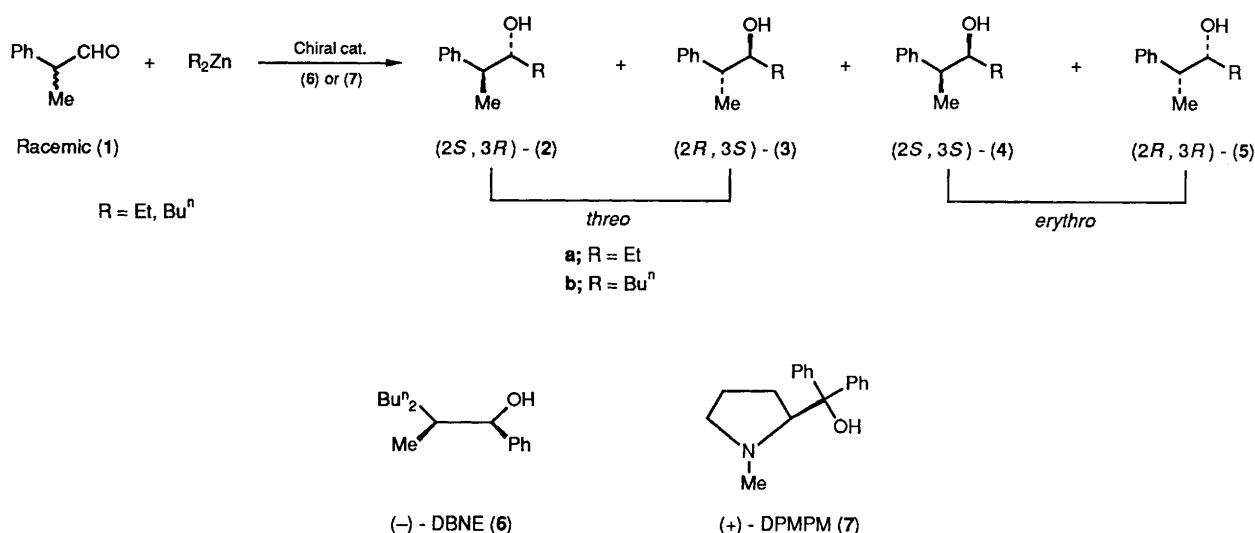


Table 1. Selective addition of R₂Zn to racemic (1) using (6) or (7) as chiral catalysts.

Entry	R	Catalyst	Yield ^a /%	Alcohols (2)—(5)			
				<i>threo</i> -[(2), (3)]		<i>erythro</i> -[(4), (5)]	
				E.e. ^b /%	Config.	E.e. ^b /%	Config.
1 ^c	Et	(6) a	60 (3.0)	93	(3a)	65	(4a)
2 ^c	Bu ⁿ	(6) b	32 (5.0)	92	(3b)	84	(4b)
3 ^d	Et	(6) a	58 (4.3)	89	(3a)	73	(4a)
4 ^e	Et	(6) a	62 ^f (6.2)	76	(3a)	76	(4a)
5 ^c	Et	(7) a	63 (6.3)	68	(3a)	24	(4a)
6 ^g	Bu ⁿ	(6) b	16 ^f (6.8)	75	(3b)	72	(4b)

^a Isolated total yield of (2)—(5). Figures in parentheses are the ratio of *erythro*-[(4) and (5)] to *threo*-[(2) and (3)] determined by HPLC analyses. ^b Determined by HPLC analyses using a chiral column [Chiralcel OD, 250 mm; 254 nm UV detector; eluant 0.5% propan-2-ol in hexane; flow rate 0.4 ml/min; column temperature 35 °C]; retention time (min) 27.9, 30.9, 33.2, 39.5 for (3a), (2a), (4a), (5a), respectively; Chiralcel OJ, 250 mm; flow rate 0.5 ml/min; column temperature 20 °C; retention time (min) 29.8, 33.7, 38.9, 43.2 for (4b), (3b), (2b), (5b). Configurations were assigned by comparison with optically active authentic samples [(2)—(5)] prepared from optically active (*S*)-(1) and RMgBr (R = Et, Buⁿ). ^c Mol ratio (1):R₂Zn:catalyst, 1:2:0.1. ^d (1):Et₂Zn:(6), 1:1:0.1. ^e (1):Et₂Zn:(6), 1:0.5:0.05. ^f Based on R₂Zn. ^g (1):Bu₂Zn:(6), 1:0.5:0.025.

2-phenylpropanal (1) without using optically active auxiliaries has been reported. However, these methods afford only racemic alcohols.⁵

We now report the first asymmetric synthesis of optically active alcohols with two chiral centres from the racemic aldehyde (1) by the addition of dialkylzinc reagents using (1*S*,2*R*)-(-)-*N,N*-dibutylnorephedrine (DBNE)^{3b,c,e,6} and (*S*)-(+)-diphenyl(1-methylpyrrolidin-2-yl)methanol (DPMPM)^{3a,c,e}, as chiral catalysts.

Reaction of racemic (1) with diethylzinc (2 equiv.) in hexane at room temperature using 10 mol% of (-)-DBNE as a chiral catalyst afforded 2-phenylpentan-3-ol in 60% yield (*threo/erythro* 1/3.0). Among the two pairs of enantiomers, (2*S*,3*R*)-*threo*-(2a) and (2*R*,3*S*)-*threo*-(3a), and (2*S*,3*S*)-*erythro*-(4a) and (2*R*,3*R*)-*erythro*-(5a), (3a) predominated for the *threo*-isomers and (4a) for the *erythro*-isomers. The enantiomeric excess (e.e.) of (3a) and (4a) reached 93 and 65%, respectively (determined by HPLC analysis using a

chiral column) (Table 1, entry 1).[†] Under slightly different conditions (1 equiv. of Et₂Zn), (3a) and (4a) were obtained in 89 and 73% e.e., respectively (entry 3).

In the reaction of Bu₂Zn with racemic-(1) using (-)-DBNE as a chiral catalyst, 2-phenylheptan-3-ol (32%; *threo/erythro* 1/5.0) was obtained. HPLC analysis showed that *threo*-(3b) and *erythro*-(4b) were formed, predominantly in 92 and 84% e.e., respectively. This selectivity of the predomi-

[†] Typical experimental procedure. Racemic (1) (0.98 mmol) was added to a solution of (-)-DBNE (0.1 mmol) in hexane (1.3 ml) at room temperature. The mixture was cooled to 0 °C, and a solution in hexane (2 ml) of Et₂Zn (2.0 mmol) was added. The mixture was stirred at room temperature for 46 h. The reaction was quenched at 0 °C by the addition of HCl (1 M; 5 ml). The resulting mixture was extracted with CH₂Cl₂ (4 × 15 ml), dried (Na₂SO₄), and evaporated under reduced pressure, and was purified by silica gel TLC (eluant AcOEt-hexane, 1:5).

nant *threo*-isomer of (**3b**) in the alkylation may be explained as follows. Because both (**3**) and (**4**) are (3*S*)-alcohols, formation of these isomers is considered to be the result of the selective addition of R₂Zn to racemic (**1**) from the *Si*-face of the aldehyde (**1**), regardless of the configuration of (**1**).[‡]

Because both enantiomers of DBNE are available, it should be possible to synthesise either enantiomer of alcohols.

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[‡] This selectivity might be termed 'pseudo-enantioselectivity.' It should be noted that the conventional diastereoselective addition affords alcohols of different configuration of C-3 depending on the configuration of (**1**).
